Acid Rain Monitoring Project

FY19 Annual Report
June 28, 2019

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Bureau of Waste Prevention

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Introduction
This report covers the period January 1, 2019 to June 30, 2019, the eighteenth year of Phase IV of the Acid Rain Monitoring Project. Phase I began in 1983 when about one thousand citizen volunteers were recruited to collect and help analyze samples from nearly half the state’s surface waters. In 1985, Phase II aimed to do the same for the rest of the streams and ponds in Massachusetts. The third phase spanned the years 1986-1993 and concentrated on a subsample of streams and ponds to document the effects of acid deposition to surface waters in the state. Over 800 sites were monitored in Phase III, with 300 citizen volunteers collecting samples and doing pH and ANC analyses. In 2001, the project was resumed on a smaller scale: about 60 volunteers are now involved to collect samples from approximately 150 sites, 26 of which are long-term sites with ion and color data dating back to Phase I. In the first years of Phase IV (2001-2003), 161 ponds were monitored for 3 years. Between Fall 2003 and Spring 2010, the project monitored 151 sites twice a year, mostly streams, except for the 26 long-terms sites that are predominantly ponds. Since 2011, reduced funding eliminated our October sampling and monitoring now occurs in April only. In 2011, we also stopped monitoring some of the streams in order to add and revisit ponds that were monitored in 2001-2003. This year is the eighth year of monitoring for those added ponds.

Goals
The goals of this project are to determine the overall trend of sensitivity to acidification in Massachusetts surface waters and whether the 1990 Clean Air Act Amendment has resulted in improved water quality.

Methods
The sampling design was changed in 2011 to monitor both streams and ponds, and that design continues to date. In 2001-2003 mostly ponds were monitored. In Fall 2003 the sampling scheme switched to streams to evaluate their response to air pollution reductions. In 2011 the site list was modified to include both ponds and streams. Half of the streams monitored since 2003 were kept, and half of the ponds monitored in 2001-2003 were added back. The streams that were removed were chosen randomly within each county. Ponds that were reinstated on the sampling list were chosen at random within those counties and by ease of accessibility to replace the removed streams. Because those sites were chosen without a preconceived plan, they can be considered picked at random.

One collection took place this year, on April 7, 2019.

Methods were unchanged from previous years: Volunteer collectors were contacted six weeks before the collection to confirm participation. Clean sample bottles were sent to them in the mail, along with sampling directions, a field sheet/chain of custody form, and directions including latitude and longitude coordinates along with maps to the sampling sites. Volunteers collected a surface water sample at their sampling sites either from the bank or wading a short distance into the water body. They collected water one foot below the surface, upstream of their body, after rinsing their sample bottle three times with pond or stream water. If collecting by a bridge, they collected upstream of the bridge unless safety and access did not allow it. They filled in their field data sheet with date, time, and site code information, placed their samples on ice in a cooler and delivered the samples to their local laboratory right away. They were instructed to collect their samples as close to the lab analysis time as possible. In a few cases, samples were collected the day prior to analysis because the lab is not open on traditional “ARM Sunday.” Previous studies by our research team have established that pH does not change significantly in 24 hours when the samples are refrigerated and stored in the dark.

One change was that Statewide Coordinator Travis Drury left UMass service, but not before sampling, and UMass analyses were completed in mid-April. He is succeeded by Cameron Richards, who took over the Statewide Coordinator position in early May.

Volunteer labs were sent any needed supplies (sulfuric acid titrating cartridge, electrode, buffers), two quality control (QC) samples, aliquot containers for long-term site samples, and a lab sheet one week to ten days before the collection. They analyzed the first QC sample (an unknown) in the week prior to the collection and called in their results to the Statewide Coordinator. If QC results were not acceptable, the volunteer analyst discussed possible reasons with the Statewide Coordinator and made modifications

1 Note: The term stream in this report refers to lotic waters (from creeks to rivers) and the term ponds refers to lentic waters (lakes and ponds, but not marshes)
until the QC sample analysis gave acceptable results. On collection day or the day after, volunteer labs analyzed the second QC sample before and after the regular samples, and reported the results on their lab sheet along with the regular samples. Analyses were done on their pH-meters with KCl-filled combination pH electrodes. Acid neutralizing capacity (ANC) was measured with a double end-point titration to pH 4.5 and 4.2. Most labs used a Hach digital titrator for the ANC determination, but some used traditional pipette titration equipment. Aliquots were taken from 24 long-term sites to fill two 50mL bottles and one 50mL tube per site for later analysis of ions and color. These aliquots were kept refrigerated until retrieved by UMass staff.

This year all 26 long-term sites were sampled, and Great Pond in Barnstable County was reinstated to the long-term sites list and analyzed for the full suite of parameters. Aliquots, empty bottles, and results were collected by the ARM Statewide Coordinator between one and three days after the collection.

The Statewide Coordinator reviewed the QC results for all labs and flagged data for any lab results that did not pass Data Quality Objectives (within 0.3 units for pH and within 3mg/L for ANC). pH and ANC data were entered by one ARM staff and proofread by another. Data were entered in a MS excel spreadsheet and uploaded into the web-based database at http://63.135.115.71/acidrainmonitoring/. Data were also posted on the ARM web page at http://wrrc.umass.edu/research/acid-rain-monitoring-project.

Water Resources Research Center's Travis Drury, with the help of sophomore student Haena Jung, managed the Environmental Analysis Lab (EAL) and provided the QC samples for pH and ANC to all of the volunteer labs. EAL also provided analysis for color analysis for the long-term site samples. The UMass Extension Soils Laboratory analyzed the samples from the long-term sites for cations, and University of New Hampshire’s Water Quality Analysis Laboratory, under the direction of Jody Potter, analyzed the samples from the long-term sites for anions.

Aliquots for 26 long-term sites were analyzed for color on a spectrophotometer within one day; anions within two months on an Ion Chromatograph; and cations within one month on an ICP at the UMass Extension Soils Laboratory on the UMass Amherst campus. The available data was sent via MS Excel spreadsheet to the Statewide Coordinator who uploaded it into the web-based database.

The Project Principal Investigator plotted the data to check for data inconsistencies and gaps. She then analyzed the available April data from 1983 through 2019, using the statistical software JMP (http://www.jmp.com/software/) to run bivariate analyses of pH, ANC, and ions against date. This yielded trends analyses with a fitted X Y line, using a 95% confidence interval.

Results

1. There were 149 sites to be monitored, 76 ponds and 73 streams. All sites were sampled this year. Of those, 19 ponds and 7 streams are “long-term” sites that are sampled every year and analyzed for color and a suite of ions in addition to pH and ANC.

2. Sampling was completed for all 149 sites (76 ponds and 73 streams) including all 26 of our long-term sites.

3. There was only one quality control problem this year, with one lab failing pH data quality objectives, affecting five sites in the northeastern part of the state.

4. For the ion analyses, both UNH and UMass analyzed the 26 long term site samples for Mg, Ca, Na, and K. This provided a comparison opportunity between the two labs, which agreed fairly well. UMass ran duplicate analyses for one site, and only one of the duplicates matched well with the UNH data, so the UMass analysis for that sample was used in the data analysis.

5. The network of volunteers was maintained and kept well informed on the condition of Massachusetts surface waters so that they would be able to participate effectively in the public debate. This was accomplished by e-mail and telephone communications, as well as through updates via an internet listserv. 56 volunteers participated in this year’s collection. Several new volunteer collectors were recruited to replace ill or retiring volunteers via Volunteermatch.org, a press release which was
picked up by at least two Massachusetts newspapers, several internet listservs, and by word of mouth. There were 10 volunteer labs across the state, in addition to the EAL at UMass Amherst, in charge of pH and ANC analyses (Table 1). As the Pepperell Waste Water Treatment lab was not available this year, their samples were analyzed by the Cushing Academy lab. The Westfield State University was short of staff, so they analyzed only half of their usual samples, and the Springfield Water and Sewer Commission analyzed the other half.

Table 1: Volunteer Laboratories

<table>
<thead>
<tr>
<th>Analyst Name</th>
<th>Affiliation</th>
<th>Town</th>
</tr>
</thead>
<tbody>
<tr>
<td>Joseph Ciccotelli</td>
<td>Ipswich Water Treatment Department</td>
<td>Ipswich</td>
</tr>
<tr>
<td>Amy Johnston</td>
<td>UMass Boston</td>
<td>Boston</td>
</tr>
<tr>
<td>Mark Putnam</td>
<td>MDC Quabbin Lab</td>
<td>Belchertown</td>
</tr>
<tr>
<td>Dave Bennett</td>
<td>Cushing Academy</td>
<td>Ashburnham</td>
</tr>
<tr>
<td>Kimberly Newton</td>
<td>Bristol Community College</td>
<td>Fall River</td>
</tr>
<tr>
<td>and Mary Rapien</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Bob Bentley</td>
<td>Analytical Balance Laboratory</td>
<td>Middleborough</td>
</tr>
<tr>
<td>Dave Christensen</td>
<td>Westfield State University</td>
<td>Westfield</td>
</tr>
<tr>
<td>Devon Avery</td>
<td>Upper Blackstone Water Pollution Abatement District</td>
<td>Millbury</td>
</tr>
<tr>
<td>Sue Tower</td>
<td>Springfield Water and Sewer Commission</td>
<td>Westfield</td>
</tr>
<tr>
<td>Cathy Wilkins</td>
<td>Greenfield High School</td>
<td>Greenfield</td>
</tr>
<tr>
<td>Travis Drury</td>
<td>UMass Amherst Environmental Analysis Lab</td>
<td>Amherst</td>
</tr>
</tbody>
</table>

6. The ARM web site and searchable database were maintained and updated. 2019 pH, ANC, color, and ion data that met data quality objectives were added to the web database via the uploading tool created in previous years. Our database consultant added components to the database to greatly improve the database download function, which resulted in a much simplified effort for our statistical analysis.

7. The data collected was analyzed for trends in pH and ANC in April months (149 sites) and for color and ions (26 sites), using the JMP® Statistical Discovery Software (http://www.jmp.com/software/). Trend analyses (scatter plots, regression, and correlation) were run on pH, ANC, color, and each ion separately for each site, predicting concentration vs. time. Note that the flagged pH data were inadvertently left in the statistical analyses for pH, but as the difference was small and didn’t affect the summary calculations, the analysis was not run again without the five affected sites.

Data Analysis Results

pH and ANC

Table 2 displays the number of sites that show a significant change over time for pH or ANC. If the difference was not statistically significant (p>0.05), the sites are tabulated in the ‘No Change’ category.

Table 2: Trend analysis results for pH and ANC, April 1983 – April 2019 (Number of sites)

<table>
<thead>
<tr>
<th></th>
<th>All sites</th>
<th>Ponds</th>
<th>Streams</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>pH</td>
<td>ANC</td>
<td>pH</td>
</tr>
<tr>
<td>Increased</td>
<td>47</td>
<td>56</td>
<td>24</td>
</tr>
<tr>
<td>Decreased</td>
<td>8</td>
<td>1</td>
<td>2</td>
</tr>
<tr>
<td>No Change</td>
<td>94</td>
<td>92</td>
<td>50</td>
</tr>
<tr>
<td>Total</td>
<td>149</td>
<td>149</td>
<td>76</td>
</tr>
</tbody>
</table>

Those results are graphed as percentages of all sites in Figure 1.
This trend analysis indicates that for most sites, neither pH nor ANC changed significantly over time. However, for those sites that show a significant change, many more show an increase than a decrease in value: 32% of the sites saw an increase in pH (30% last year) and 38% had an increase in ANC (33% last year). It can be noted that a larger percentage of sites have seen an increase in pH and ANC this year than last year.

Again this year we see a much higher percentage of ponds exhibiting an increase in ANC compared to streams (49% vs. 26%), but unlike last year when more streams had a higher pH than in the past than ponds (32% vs. 29%), this year both streams and ponds have 32% of water bodies showing an increase in pH. As far as decreases in pH are concerned, the situation is the same as last year: more streams saw a drop in pH (8%) than ponds (3%), while only 1% of streams and no ponds saw a decrease in ANC.

The 2018-2019 winter saw less snowfall than in the past couple of years. We purposely sample in early April to catch any large snowmelt events, but this year there was no snow left on the ground by April 7, and our results indeed do not indicate an acid pulse due to snowmelt.

**Ions and color**

Trend analyses were run for 26 long-term sites that were analyzed for eleven ions and for color. Results are shown in Table 3 and Figure 2. Note that the trends period is 1985-2019.
Table 3: Trends for number of sites with increases or decreases in ion concentration and color April 1985 – April 2019

<table>
<thead>
<tr>
<th></th>
<th>Increased</th>
<th>Decreased</th>
<th>No Change</th>
</tr>
</thead>
<tbody>
<tr>
<td>Color</td>
<td>22</td>
<td>1</td>
<td>3</td>
</tr>
<tr>
<td>Cl</td>
<td>17</td>
<td>0</td>
<td>9</td>
</tr>
<tr>
<td>NO3</td>
<td>10</td>
<td>1</td>
<td>15</td>
</tr>
<tr>
<td>SO4</td>
<td>0</td>
<td>24</td>
<td>2</td>
</tr>
<tr>
<td>Mg</td>
<td>6</td>
<td>0</td>
<td>20</td>
</tr>
<tr>
<td>Mn</td>
<td>0</td>
<td>9</td>
<td>17</td>
</tr>
<tr>
<td>Fe</td>
<td>0</td>
<td>6</td>
<td>20</td>
</tr>
<tr>
<td>Cu</td>
<td>1</td>
<td>0</td>
<td>25</td>
</tr>
<tr>
<td>Al</td>
<td>1</td>
<td>4</td>
<td>21</td>
</tr>
<tr>
<td>Ca</td>
<td>4</td>
<td>1</td>
<td>21</td>
</tr>
<tr>
<td>Na</td>
<td>14</td>
<td>0</td>
<td>12</td>
</tr>
<tr>
<td>K</td>
<td>8</td>
<td>0</td>
<td>18</td>
</tr>
</tbody>
</table>

Figure 2: Results of trend analysis for ions and color at 26 long-term sites, April 1985-2019
Shown is how many sites showed an increase (blue), decrease (orange), or no significant change (yellow) over the period 1985 – 2019

Results are similar to previous years. While there are still more sites that show no significant change either up or down, more cations display an increase than a decrease over the years. Sodium continues to be the ion with the most increases, with manganese showing the most decreases.

For anions, we continue to see a very significant downward trend in Sulfate (24 sites). Nitrates, on the other hand, continue to show more increases than decreases, and it is unknown at this time whether it is due to increasing vehicular emissions, or a result of climate change – smaller and less persistent snowpacks result in fine root damage and reduced microbial activity. This can result in losses of nutrient elements, most notably Nitrogen in the form of NO₃⁻.

Color is still increasing in most of our sites, which is consistent with a recovery of natural alkalinity.
Discussion

The continued trend in decreasing sulfate confirms that the Clean Air Amendment of 1990 is having a positive effect in the quality of the Commonwealth’s surface water quality. Road salting in the winter continues to affect the concentration of sodium and calcium in the water bodies. Continued monitoring will help tease out whether nitrate pollution is countering the beneficial effect of decreased sulfates.

Acknowledgements

Thank you to all of the project’s volunteers who make this project possible by collecting samples all over the state under any weather conditions, and who spend many hours in the lab analyzing samples.

Cover photo: Volunteers take a sample on April 7, 2019.
Appendix

Table A-1: April 7, 2019 pH and Alkalinity data
Note: Values in red did not pass pH quality control.

<table>
<thead>
<tr>
<th>Name</th>
<th>Palsite</th>
<th>pH</th>
<th>Alkalinity</th>
</tr>
</thead>
<tbody>
<tr>
<td>Aldrich Brook</td>
<td>5131425</td>
<td>6.26</td>
<td>5.8</td>
</tr>
<tr>
<td>Angeline Brook</td>
<td>9560000</td>
<td>5.51</td>
<td>1.6</td>
</tr>
<tr>
<td>Anthony Brook</td>
<td>2105425</td>
<td>6.49</td>
<td>5.7</td>
</tr>
<tr>
<td>Ashby Reservoir</td>
<td>81001</td>
<td>6.45</td>
<td>2.5</td>
</tr>
<tr>
<td>Ashfield Lake</td>
<td>33001</td>
<td>7.23</td>
<td>40.1</td>
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<td>Barton Brook</td>
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<td>6235800</td>
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<td>21010</td>
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<td>11002</td>
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<td>55.8</td>
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<td>51024</td>
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<td>Location</td>
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<td>Alkalinity</td>
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Data in red characters are flagged due to the laboratory failing the pH QC test.
Table A-2: April 7, 2019 color and ion concentration for 26 long term sites. Ion concentrations are in mg/L, color in PCU

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